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Electrochemistry Multiple Choice Questions Answers

this guide covers electrochemistry multiple choice questions (mcqs) with answers and explanation.

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Electrochemistry Practice Test with Answers. Question 1. The standard EMF for the given cell reaction $Zn + Cu^{2+} = Cu + Zn^{2+}$ is 1.10 V at 25 o C. The EMF for the cell reaction, when 0.1 M

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Cu 2+ and 0.1 M Zn 2+ solutions are used, at 25 o C is. Related: Question Bank for NEET Physics.

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AP Chemistry: Electrochemistry Multiple Choice Answers 14. Questions 14-17 The spontaneous reaction that occurs when the cell in the picture operates is as follows: $2\text{Ag}^+ + \text{Cd (s)} \rightarrow 2\text{Ag (s)} + \text{Cd}^{2+}$ (A) Voltage increases. (B) Voltage decreases but remains $>$ zero.

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Multiple Choice Questions 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the _____ ; the chemical change that occurs at this electrode is called _____.

Sample Questions - Chapter 21

APEF - Electrochemistry - Multiple Choice Questions A 1. Which of these chemical reactions is an oxidation-reduction reaction? A. $\text{Fe} + \text{S} \hat{=} \text{FeS}$ C. $\text{AgNO}_3 + \text{NaCl} \hat{=} \text{AgCl} + \text{NaNO}_3$ B. $\text{CO}_2 + \text{H}_2\text{O} \hat{=} \text{H}_2\text{CO}_3$ D. $\text{H}_2\text{SO}_4 + 2\text{NaOH} \hat{=} \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$ C 2. What happens to the oxidizing agent in an oxidation-reduction reaction? A. It is oxidized as it gains electrons. C.

APEF - Electrochemistry - Multiple Choice Questions ...

Multiple Choice Questions (Type-II) Note : In the following questions two or more than two options may be correct. The

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positive value of the standard electrode potential of $\text{Cu}^{2+} / \text{Cu}$ indicates that _____. (i) this redox couple is a stronger reducing agent than the H^+ / H_2 couple. (ii) this redox couple is a stronger oxidising agent than H^+ / H_2 .

Class 12 Important Questions for Chemistry - Electrochemistry

Sample Questions - Chapter 21 Electrochemistry Examples of Multiple Choice Questions: 1. In an electrolytic cell the electrode at which the electrons enter the solution is called the ____; the chemical change that occurs at this electrode is called _____. (a) anode, oxidation (b) anode, reduction (c) cathode, oxidation (d) cathode, reduction

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SAHOTA 03 Electrochemistry Study Guide - Multiple Choice - Page 1 of 22 1. DO ALL. THE. QUESTIONS in this booklet. These are actual Provincial Exam questions! Your own provincial exam and unit test will include questions similar to the ones in this booklet! 2. RESIST THE. URGE

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Choice; ... Wrong Answer Wrong Explanation Wrong Question
Question not related to topic Spelling Mistakes. Ask a Doubt.

NEET Chemistry Electrochemistry Questions Solved

31. During electrolysis of water the volume of oxygen liberated is 2.24 dm^3 . The volume of hydrogen liberated, under same conditions will be : 2.24 dm^3 . 1.12 dm^3 . 4.48 dm^3 . 0.56 dm^3 .
Answer. 32. In Down's method for the extraction of sodium, the melting point of the electrolyte is lowered by adding.

Important Questions of Electrochemistry for NEET Chemistry ...

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C.E. Housecroft and E.C. Constable. Once you have answered the questions, click on ...

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AP Chemistry: Electrochemistry Multiple-Choice Practice

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MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) The gain of electrons by an element is called _____. A) oxidation B) sublimation C) reduction D) disproportionation E) fractionation 2) _____ is reduced in the following reaction: $\text{Cr}_2\text{O}_7^{2-} + 6\text{S}_2\text{O}_3^{2-} + 14\text{H}^+ \rightarrow 2\text{Cr}^{3+} + 3\text{S}_4\text{O}_6^{2-} + 7\text{H}_2\text{O}$

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